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Background

Next-Gen Surface Ligands for Next-Gen Technologies Raman Scattering: The inelastic scattering of light, most often attributed to molecular vibrations. It is Electrochemical aptamer-based sensors (E-ABs) utilizes however a very weak effect in that only roughly 1 in aptamer as the recognition element, enables, continuous, every 10⁶ scattered photons is inelastically real time, and *in vivo* monitoring of target. scattered V Thiols are Target binds to aptamer, changes its shape and causes shift in A displaced the electrochemical signal (figure, left) that can be measured. Target = (Hyper-Raman Scattering: The non-linear analog of 10 20 E-AB technology utilizes gold-thiol (Au-S) bond and the Time (hours) Raman scattering where two photons of incident stability of the monolayer is limited under biological and Competitive Displace frequency ω_0 give way to a scattered photon of electrochemical conditions. Figure (left) shows NHCs are not frequency ω_{HR} that is shifted relative to the second displaced by cysteine *via* competitive displacement harmonic frequency ($\omega_{HR} = 2\omega_0 \pm \omega_{vib}$). This effect NHCs are not V displaced! is even weaker than normal Raman scattering. In Situ Spectroelectrochemistry No Competitive Displacement During Cyclic Voltammetry: 0.2 V to -0.5 V along a metal/dielectric interface, can be viewed as oscillations of the conduction electrons Method • Deposition protocol (left) matters for the -Before quality of monolayer is crucial for EAB stability KHMDS and performance Electric Electron -100 cycles A systematic comparison of various widely $-\omega_{vib}$ used deposition protocols show that the Eloc fundamental characteristics of the Au-NHC 120 °C, 24 min 55 °C, 24 hours monolayer is heavily dependent on the -600 cycles Method 2 Method 3 deposition protocol As Method 2 produced high quality -5100 cycles Metal nanostructure monolayer, the same has been employed to nanoparticle - M.W. study the stability of monolayer upon -11250 cycles continuous voltametric cycling (figure, right). Surface Enhanced Raman Scattering (SERS): Adsorption of a molecule on a metal surface 900 1100 1300 1500 1700 Raman Shift (cm⁻¹) hyper-Raman scattering (SEHRS). Chandran, Dominique, Kaur, Clark, Nalaoh, Ekowo, Jensen, Dominique, Chandran, Jensen, Jenkins, Camden Aloisio, Crudden, Arroyo-Curras, Jenkins, Camden. Nanoscale. Chem. A European Journal. 2023. 15, e202303681 **Room Temperature Methanolic Deposition** accounts for 10⁴ (SERS) and 10⁶ (SEHRS). NHC ligands as a Mass Ionization Tag for Biomolecules in LDI-MS SEHRS and the Chemical Effect Carbodiimide coupling scheme for peptide immobilization -9-AL Surface-induced [(2)2Au] The other contributor is the chemical EDC/NHS, PBS 9-Au. 10-Au [(2)(9)Au] 1831.35 mechanism (left) 10. R = L-Lvsin 9. R = L-Lysine methyl este [(9)₂Au which has been 973.48 *approximated* to be and the second se around the order of -10-Au · LNAN-10¹-10² (SERS) and Conventional fragment ligands **Chemical Mechanism** 10²-10⁴ (SEHRS). E) Charge Transfer (CT) extensively in LDI-MS. C) Static Chemical (CHEM) Molecular Resonance (RS) • NHC ligands enable the desorption/ionization of biomolecules without fragmentation. the extremely low intensity HR signal of non-surface-enhanced molecules and the difficulty Carboxylic acid NHC was used to capture Llysine methyl ester (Figure, left) and LVTDLTKof disentangling the chemical mechanism contributions (CHEM, RS, CT). 800 750 850 OMe peptide (Figure, right) m/zm/z - The string of wavelengths, we can see Fundamental Vibrational Spectroscopy Studies of NHC on Gold Nanoparticles if a change in the SEHRS spectrum is observed that We compared the experimental Raman, SERS and IR spectra of the diisopropyl benzimidazolium NHC can arise from: Molecular resonanceand its ¹³C-labeled isotopologue to first principles theory to definitively assign previously unreported vibrational modes and verify and compile normal modes from prior literature. hyper-Raman (RS) \succ The creation of new metal charge transfer SERS active normal mode Experiment states (CT). -----¹²C-AuNP ¹³C-AuNE



Chandran, Camden. ACS Nano. 2024. 18. 20827-20834 **Localized Surface Plasmon Resonance (LSPR):** Surface electromagnetic waves that propagate (left). Excitation of the plasmon provides the local field enhancement required for SERS (right). increases the Raman effect by $10^6 - 10^8$. The two-photon analog of SERS is surface-enhanced The primary contributor to the overall enhancement is the electromagnetic mechanism, which • However, determining SEHRS enhancement factors remains a challenging problem due to • For R6G (right-1st panel),



- By scanning through a series
- the laser is scanned through excited first state its $(S_1 \leftarrow S_0, 2\omega = 515 \text{ nm}).$
- > Dramatic changes in the relative intensities of modes coupled with the S_1 excited state are observed (gray shaded box), suggesting the presence of resonance effects.



The wavelength-scanned spectra of p-MPN (above right) are also presented, which remain consistent across all wavelengths.

> This suggests that there are no molecular resonancehyper-Raman effects, at least in this wavelength range.

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With isotope labeling, we also isolate the could normal modes with Au-C vibration, which allows us to determine chemisorption of the NHC on gold using a few peaks in the SERS spectra.

Chowdhury, Hu, Jenkins, Jensen, Camden, et al. J. Phys. Chem. C. 2024, 128, 32, 13550 Jensen, Chowdhury, Jenkins et al. Chem. Commun. 2023, 59, 13524



Using the world class electron microscope available to us at Oak Ridge National Laboratory, we have recently characterized the near-field response of individual gold nanotriangles (NT) over a broad, visible-toinfrared spectral region. EEL spectrum images for a 1420 nm gold NT displaying the spatial profiles of its m=1-7 Fabry-Pérot modes (a), and the EEL spectra of a set of gold NTs vs edge length (b) is shown on the right.





A scanning transmission electron microscope (STEM) can probe localized surface plasmons with nanometer spatial resolution and ultrahigh energy resolution using electron energy loss spectroscopy (EELS)

